B. Sc. IInd Year (IIIrd Semester) Paper-X (CH-203) Organic Chemistry(Theory) M. Marks: 32 Time: 3 Hrs.

Note: Nine questions will be set. Q. No. 1, based on whole syllabus, is compulsory. There will be four questions from section A and four from section **B**. Candidates will be required to attempt five questions in all, selecting at least two questions from each section. Question no. 1 carry 8 marks and all questions in Section A & B (not more than 2 - 3 parts) carry 6 marks each.

Section-A (23 Periods)

Alcohols

Monohydric alcohols — nomenclature, methods of formation by reduction of aldehydes, ketones, carboxylic acids and esters. Hydrogen bonding. Acidic nature. Reactions of alcohols.

Dihydric alcohols — nomenclature, methods of formation, chemical reactions of vicinal glycols, oxidative cleavage $[Pb(OAc)_4]$ and HIO₄] and pinacol-pinacolone rearrangement.

Phenols

Nomenclature, structure and bonding. Preparation of phenols, physical properties and acidic character. Comparative acidic strengths of alcohols and phenols, resonance stabilization of phenoxide ion. Reactions of phenols — electrophilic aromatic substitution, Mechanisms of Fries rearrangement, Claisen rearrangement, Reimer- Tiemann reaction, Kolbe's reaction and Schotten and Baumann reactions.

Epoxides

Synthesis of epoxides. Acid and base- catalyzed ring opening of epoxides, orientation of epoxide ring opening, reactions of Grignard and organolithium reagents with epoxides.

Section-B (22 Periods)

Ultraviolet (UV) absorption spectroscopy

Absorption laws (Beer- Lambert law), molar absorptivity, presentation and analysis of UV spectra, types of electronic transitions, effect of conjugation. Concept of chromophore and auxochrome. Bathochromic, hypsochromic, hyperchromic and hypochromic shifts. UV spectra of conjugated enes and enones, Woodward- Fieser rules, calculation of λ_m _{a x} of simple conjugated dienes and α,β - unsaturated ketones.

Applications of UV Spectroscopy in structure elucidation of simple organic compounds.

Carboxylic Acids & Acid Derivatives

Nomenclature of Carboxylic acids, structure and bonding, physical properties, acidity of carboxylic acids, effects of substituents on acid strength. Preparation of carboxylic acids. Reactions of carboxylic acids. Hell- Volhard- Zelinsky reaction. Reduction of carboxylic acids. Mechanism of decarboxylation.

Relative stability of acyl derivatives. Physical properties, interconversion of acid derivatives by nucleophilic acyl substitution. Mechanisms of esterification and hydrolysis (acidic and basic)